ATOINS

WHAT IS AN ATOM?

- Atoms are the building blocks of all matter
- EVERYTHING is made up of <u>different</u>
 <u>combinations</u> of atoms

 Atoms are made up of a nucleus (at its center) and an electron cloud (surrounding the nucleus)

SUBATOMIC PARTICLES

- Sub= under (think <u>sub</u>marine)
- Subatomic particles are the smaller parts that make up an atom
- Atoms are made up of 3 different subatomic particles
 - Protons
 - Neutons
 - Electrons

THE THREE SUBATOMIC PARTICLES

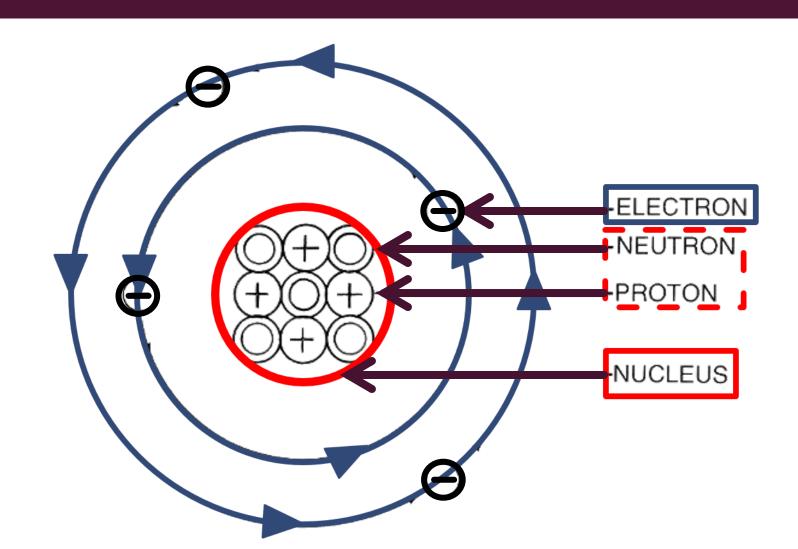
- Protons- positively charged (+), found IN the nucleus; mass of I atomic mass unit (amu)
- Neutrons- no charge (0), found in the nucleus; mass of lamu
- Electrons- negatively charged (-), found in the electron cloud, much smaller than protons and neutrons (mass << I amu)</p>

ELECTRON CLOUD THEORY

Modern Atomic Theory (the theory we use to explain the structure of atoms) is called <u>electron cloud theory</u>

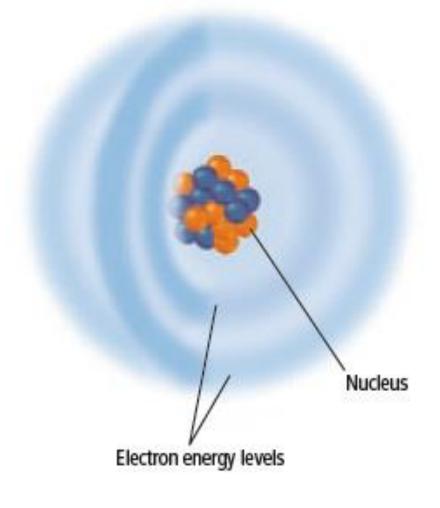
Says that an atom has a nucleus surrounded by an "electron cloud" that has different energy levels (layers) and that each can hold a different number of electrons and electrons fill layers closest to the nucleus first

WHAT WE THINK ATOMS LOOK LIKE TODAY



WHY IS IT AN ELECTRON "CLOUD"?

Electrons orbit the nucleus somewhere in their energy level- they don't have an exact spot like planets orbiting the sun but instead move around their level



VALENCE ELECTRONS

- Valence electrons are found in the outermost energy level of the electron cloud- they're on the outside of the cloud
- These are the **ONLY** electrons involved in bonding

- Valence electrons determine...
 - The chemical properties of an element
 - Reactivity- the ability of an atom to chemically react with another atom

VALENCE ELECTRON PAIRINGS

■ 8 is the magic number- if there are 8 valence electrons (electrons in the outermost energy level) the atom is stable and doesn't want to react with other atoms

Hydrogen (H), helium (He), lithium (Li), and beryllium (Be) are the exceptions. They only need <u>2</u> electrons to be stable

WHERE THE VALENCE SHELL IS

